Molecular Hydrogen as a Novel Antioxidant: Overview of the Advantages of Hydrogen for Medical Applications

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Contents

1.	Introduction		290
2.	. Comparison of H ₂ with Other Medical Gasses		291
3.	. Oxidative Stress as Pathogenic Sources		292
4.	. Physiological Roles of H ₂ O ₂		294
5.	Measurement of H ₂ Gas Concentration		295
6.	Advantages of Hydrogen in Medical Applications		296
	6.1	Selective reaction of H ₂ with highly reactive ROS	296
	6.2	Rapid diffusion	298
7.	Methods of Ingesting Molecular Hydrogen		301
	7.1	Inhalation of hydrogen gas	301
	7.2	Oral ingestion by drinking hydrogen water	301
	7.3	Injection of hydrogen-saline	302
	7.4	Direct incorporation of molecular hydrogen by diffusion: Eye drops,	
		bath, and cosmetics	304
	7.5	Maternal intake of H ₂	304
8.	Medical Effects of H ₂		304
	8.1	Acute oxidative stress by ischemia/reperfusion	304
	8.2	Chronic oxidative stress loading to neurodegeneration	305
	8.3	Stimulatory effects on energy metabolism	305
	8.4	Anti-inflammatory effects	306
9.	Possible Molecular Mechanisms Underlying Various Effects of Molecular		
	Hydrogen		306
	9.1	Direct reduction of hydroxyl radicals with molecular hydrogen	306
	9.2	Direct reduction of peroxynitrite with molecular hydrogen to regulate gene	
		expression	307
	9.3	Indirect reduction of oxidative stress by regulating gene expression	308
١0.	Unr	esolved Questions and Closing Remarks	309
Refe	eferences		

Abstract

Molecular hydrogen (H₂) was believed to be inert and nonfunctional in mammalian cells. We overturned this concept by demonstrating that H₂ reacts with highly reactive oxidants such as hydroxyl radical (*OH) and peroxynitrite (ONOO⁻) inside cells. H₂ has several advantages exhibiting marked effects for medical applications: it is mild enough neither to disturb metabolic redox reactions nor to affect signaling by reactive oxygen species. Therefore, it should have no or little adverse effects. H₂ can be monitored with an H₂-specific electrode or by gas chromatography. H₂ rapidly diffuses into tissues and cells to exhibit efficient effects. Thus, we proposed the potential of H₂ for preventive and therapeutic applications. There are several methods to ingest or consume H₂: inhaling H₂ gas, drinking H₂-dissolved water (H₂-water), injecting H₂-dissolved saline (H₂-saline), taking an H₂ bath, or dropping H₂-saline onto the eyes. Recent publications revealed that, in addition to the direct neutralization of highly reactive oxidants, H2 indirectly reduces oxidative stress by regulating the expression of various genes. Moreover, by regulating gene expression, H₂ functions as an anti-inflammatory, antiallergic, and antiapoptotic molecule, and stimulates energy metabolism. In addition to growing evidence obtained by model animal experiments, extensive clinical examinations were performed or are under way. Since most drugs specifically act on their specific targets, H₂ seems to differ from conventional pharmaceutical drugs. Owing to its great efficacy and lack of adverse effects, H_2 has potential for clinical applications for many diseases.

1. INTRODUCTION

Molecular hydrogen with the molecular formula H₂ is a colorless, odorless, tasteless, nonmetallic, and nontoxic gas at room temperature. Hydrogen gas is flammable and will burn in air at a very wide range of concentrations between 4% and 75% by volume. Its autoignition temperature, the temperature of spontaneous ignition in air, is about 500 °C (http://en.wikipedia.org/wiki/Hydrogen). These facts suggest that H₂ is not so dangerous in daily life when its concentration is under 4%.

In turns of biological reactions in several microorganisms, H_2 is a product of certain types of anaerobic metabolism, usually via reactions catalyzed by iron- or nickel-containing enzymes called hydrogenases (Adams, Mortenson, & Chen, 1980; Fritsch, Lenz, & Friedrich, 2013). H_2 is also enzymatically metabolized as an energy source by providing electrons to the electron transport chain. These enzymes catalyze the reversible redox reaction between H_2 and its constituent two protons and two electrons (van Berkel-Arts et al., 1986).

On the other hand, in all photosynthetic organisms, the water-splitting reaction occurs in the light reactions, where water is decomposed into

protons, electrons, and oxygen. Some organisms, including the alga *Chlamydomonas reinhardtii* and cyanobacteria, have evolved a second step in the dark reactions in which protons and electrons are reduced to form H₂ gas by specialized hydrogenases in cyanobacteria or chloroplast (Carrieri, Wawrousek, Eckert, Yu, & Maness, 2011). For industrial uses, extensive efforts have also been undertaken with alga in a bioreactor by genetically modifying cyanobacterial hydrogenases to synthesize H₂ gas efficiently (King, 2013; van Berkel-Arts et al., 1986).

In contrast, H₂ was accepted to behave as an inert gas in mammalian cells because of the lack of no hydrogenase genes. Thus, it had been believed that H₂ is nonfunctional in our cells. In fact, H₂ seemed to react with no biological compounds, including oxygen (O₂), in the absence of catalysts at body temperature. Indeed, owing to its characteristics, H₂ gas was used for measuring local blood flow (Aukland, Bower, & Berliner, 1964).

We overturned this concept in a publication in 2007 describing that H₂ acts as a therapeutic and preventive antioxidant by selectively reducing highly active oxidants, such as hydroxyl radical (*OH) and peroxynitrite (ONOO⁻) in cultured cells, and that H₂ has cytoprotective effects against oxidative stress (Ohsawa et al., 2007). Since then, a large number of studies have explored therapeutic and preventive effects of H₂. These publications cover many biological effects against oxidative stress in almost all organs (Ohta, 2011, 2012). Moreover, it has been revealed that H₂ has more roles, including anti-inflammatory, antiapoptotic, and antiallergic effects, in most tissues of model animals, and that H₂ stimulates energy metabolism. In addition to publications on model animal experiments, more than 10 papers on clinical examinations have been published. As of 2013, the number of publications on its biologically or medically beneficial effects had surpassed 300 (Ohta, 2014).

2. COMPARISON OF H₂ WITH OTHER MEDICAL GASSES

Gas inhalation as disease therapy has recently received attention (Kajimura, Fukuda, Bateman, Yamamoto, & Suematsu, 2010; Szabó, 2007). In recent decades, there has been extraordinary and rapid growth in our knowledge of gaseous molecules, including hydrogen sulfide (H₂S), nitric oxide (NO[•]), and carbon monoxide (CO). H₂S, CO, and NO[•] are extremely toxic molecules; however, they play important roles as signaling molecules in biological systems (Kimura, 2010; Motterlini & Otterbein, 2010).

In contrast, H₂ has advantages in terms of toxicity: it has no cytotoxicity even at high concentration (Abraini, Gardette-Chauffour, Martinez, Rostain, & Lemaire, 1994; Fontanari et al., 2000; Lillo & Parker, 2000; Lillo, Parker, & Porter, 1997). Furthermore, safety standards have been established for high concentrations of hydrogen gas for inhalation since high-pressure hydrogen gas was actually used in deep diving gas mixes to prevent decompression sickness and arterial gas thrombi (Fontanari et al., 2000). The safety of H₂ for humans is demonstrated by its application in Hydreliox, an exotic, breathing gas mixture of 49% H₂, 50% helium, and 1% O₂, which is used to prevent decompression sickness and nitrogen narcosis during very deep technical diving (Abraini et al., 1994; Fontanari et al., 2000; Lillo & Parker, 2000; Lillo et al., 1997).

As the primary target of H_2S , CO, and NO^{\bullet} , heme-based proteins play central roles. Integrated approaches revealed the physiological significance of H_2S , CO, and NO^{\bullet} on mitochondrial cytochrome c oxidase, a key target and central mediator of mitochondrial respiration (Kajimura et al., 2010). As far as briefly examined (Ohsawa et al., 2007), H_2 does not reduce the oxidized heme of cytochrome c. Thus, the primary target of H_2 seems to differ from that of the other medical gaseous molecules.

Moreover, the production of NO $^{\bullet}$, H₂S, or CO is carried out by different enzymes, NO $^{\bullet}$ synthases, cystathionine γ -lyase/cystathionine β -synthase, or hemeoxygenase-1 (HO-1), respectively (Kashfi & Olson, 2013). In contrast, as mentioned above, mammalian cells have no enzyme for producing intracellular H₂.

Regarding the interaction between H₂ and the other toxic medical gasses, combined therapy with H₂ and CO demonstrated additional therapeutic efficacy via both antioxidant and anti-inflammatory mechanisms, and may be a clinically feasible approach for preventing ischemia/reperfusion injury in the myocardium (Nakao et al., 2010). Breathing NO plus H₂ during ischemia/reperfusion reduced the infarct size and maintained cardiac function, and reduced the generation of myocardial nitro-tyrosine associated with NO inhalation (Shinbo et al., 2013). These findings suggest that the target of H₂ differs from those of CO and NO.

3. OXIDATIVE STRESS AS PATHOGENIC SOURCES

First, the author would like to introduce how the biological function of H₂ was discovered regarding its contribution to reducing oxidative stress.

Reactive oxygen species (ROS) are generated inside the body during daily life as a by-product of energy metabolism by oxidative phosphorylation in every aerobic organism. Occasionally, excess ROS are produced, such as by smoking or air pollution, exposure to ultraviolet or irradiation rays, intense exercise, and physical or psychological stress (Agarwal, 2005; Grassi et al., 2010; Harma, Harma, & Erel, 2006; Liu et al., 1996; Tanriverdi et al., 2006). When ROS are produced excessively or endogenous antioxidant capacity is diminished, indiscriminate oxidation elicits harmful effects, resulting in "oxidative stress."

Acute oxidative stress arises from various different situations: inflammation, ischemia/reperfusion in cardiac or cerebral infarction, organ transplantation, and cessation of operative bleeding, among others (Ferrari et al., 1991; Reuter, Gupta, Chaturvedi, & Aggarwal, 2010; Vaziri & Rodriguez-Iturbe, 2006). Under normal conditions, ROS induced by strenuous exercise result in muscle fatigue (Westerblad & Allen, 2011). Evidence has established strong links between chronic oxidative stress and a wide variety of pathologies, including malignant diseases, diabetes mellitus, atherosclerosis, and chronic inflammatory processes, as well as many neuro-degenerative diseases and the aging process (Andersen, 2004; El Assar, Angulo, & Rodriguez-Manas, 2013; Kim & Byzova, 2014).

As a first step in generating ROS, superoxide anion radicals (${}^{\bullet}O_2^{-}$) are the primary ROS mostly generated by electron leakage from the mitochondrial electron transport chain (Andersen, 2004; Finkel & Holbrook, 2000; Lin & Beal, 2006; Turrens, 2003). Other enzymes, including NADPH oxidases, cytochrome p450s, lipoxygenase, cyclooxygenase, and xanthine oxidase, also participate in ROS generation in the immune- or detoxifying system (Droge, 2002). Superoxide dismutase enzymatically converts ${}^{\bullet}O_2^-$ to hydrogen peroxide (H₂O₂), which is metabolized to generate water (H₂O). Highly reactive OH is generated from H₂O₂ or O₂ via the Fenton or Weiss reaction in the presence of catalytically active metals, such as Fe²⁺ and Cu⁺ (Halliwell & Gutteridge, 1992). Reaction of O₂ with NO generates ONOO, which is a very active nitrogen species (RNS) (Radi, 2013). OH is the major cause of the oxidation and destruction of biomolecules by direct reaction or by triggering the chain reaction of free radicals (Lipinski, 2011). Ionizing radiation, including cosmic rays, also generates OH as a damaging intermediate through the reaction with water, a process termed radiolysis (Schoenfeld, Ansari, Nakao, & Wink, 2012; Schoenfeld et al., 2011).

Although antioxidation therapy or prevention of various diseases is expected owing to the clinical importance of oxidative damage, many

antioxidants have been of limited therapeutic success (Steinhubl, 2008). Antioxidant supplements have exhibited little effect on preventing cancer, myocardial infarction, and atherosclerosis, but conversely have increased mortality (Bjelakovic, Nikolova, Gluud, Simonetti, & Gluud, 2007; Brambilla et al., 2008; Hackam, 2007; Hercberg et al., 2010; Steinhubl, 2008).

4. PHYSIOLOGICAL ROLES OF H₂O₂

As mentioned above, ROS had historically been believed to cause cellular damage and to lack physiological functions; however, cellular redox homeostasis is a delicate balance between ROS production and the antioxidant system (Bashan, Kovsan, Kachko, Ovadia, & Rudich, 2009; Brewer, Mustafi, Murray, Rajasekaran, & Benjamin, 2013). Some ROS are now appreciated to function as signaling molecules to regulate a wide variety of physiological process (Bell, Klimova, Eisenbart, Schumacker, & Chandel, 2007; Liu, Colavitti, Rovira, & Finkel, 2005). H₂O₂ was shown to be required for cytokine, insulin, growth factor, AP-1, c-Jun N-terminal kinase 1, p53, and nuclear factor kappa B signaling and to promote phosphatase inactivation by cysteine oxidation (Chandel, Trzyna, McClintock, & Schumacker, 2000; Chandel, Vander Heiden, Thompson, Schumacker, 2000; Finkel, 1998). These reactions provide a plausible biochemical mechanism by which ROS can impinge on signaling pathways (Collins et al., 2012).

Additionally, oxidative stress caused by H₂O₂ and NO induces enzymes involved in antioxidation and tolerance to protect cells against oxidative stress (Endo et al., 2009; Ristow & Zarse, 2010). For example, translocation of NF-E2-related factor 2 (Nrf2) into the nucleus leads to the regulation of gene expression involved in defense systems against oxidative stress (Jazwa & Cuadrado, 2010) and other toxic sources including heavy metals (Gan & Johnson, 2014). Moreover, H₂O₂ is a key factor to regulate cellular differentiation (Tormos et al., 2011; Tsukagoshi, Busch, & Benfey, 2010), the immune system (West et al., 2011; Zhou, Yazdi, Menu, & Tschopp, 2011), autophagy (Garg et al., 2013; Li, Ishdorj, & Gibson, 2012), and apoptosis (Mates, Segura, Alonso, & Marquez, 2012). Thus, it is crucial for functional H₂O₂ not to be completely eliminated in order to maintain homeostasis; as such, it is very important to be aware of side effects when developing an effective antioxidant for the prevention of oxidative stress-related diseases.

Unexpectedly, recent notable studies have suggested that excessive antioxidants increased mortality and rates of cancer (Bjelakovic et al., 2007; Bjelakovic, Nikolova, Gluud, Simonetti, & Gluud, 2008; Gray et al., 2008; Hackam, 2007; Hercberg et al., 2010; Walker, 2008) probably because they may interfere with some essential defensive mechanisms (Bjelakovic & Gluud, 2007; Bjelakovic et al., 2008; Carriere et al., 2004; Chandel et al., 1998; Mandal et al., 2010; Miller et al., 2005; Salganik, 2001). Against this background, an ideal antioxidant is expected to mitigate excess oxidative stress, but not disturb redox homeostasis. In other words, an ideal molecule should not reduce signaling molecules, such as H₂O₂ but should effectively reduce strong oxidants, such as OH.

Since H_2 reduces OH but does not react with O_2 , H_2O_2 , and NO that have physiological roles (Ohsawa et al., 2007), we propose that the adverse effects of H_2 are very small compared with those of other antioxidants. Thus, we have reached the conclusion that the ideal antioxidant could be H_2 .

5. MEASUREMENT OF H₂ GAS CONCENTRATION

H₂ gas concentration is measureable by gas chromatography. Additionally, H₂ concentration dissolved in a solution can be measured by this method. For example, H₂ in blood can be monitored by the following method: Venous or arterial blood (e.g., 5 ml) is collected in a closed aluminum bag with no dead space, followed by the addition of a defined volume of air (e.g., 30 ml) into the bag. After complete transfer of the H₂ gas from the blood to the air in the closed bag, H₂ can be measured by gas chromatography (Fig. 1). The inhalation of H₂ actually increased H₂ dissolved in arterial blood in a hydrogen gas concentration-dependent manner, and the H₂ levels in venous blood were lower than in arterial blood; the different level between arterial and venous blood indicates the amount of H₂ incorporated into and consumed by tissues (Ohsawa et al., 2007). In a clinical examination, Ono et al. also showed a difference in H₂ concentrations between arterial and venous blood (Ono et al., 2012).

 H_2 concentration can be measured using an H_2 electrode that specifically detects H_2 ; however, this sensor is also somewhat sensitive to H_2S . Thus, when H_2S is contaminated in a solution, one must take into consideration its effects.

H₂ can be measured in tissues using a needle-type H₂ sensor (Unisense, Aarhus, Denmark). The electrode current was measured with a picoammeter (Keithley, Cleveland, Ohio) attached to a recorder. The negative



- 1 +Blood
- 2 +Air3 Gas chromatography

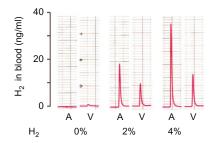


Figure 1 Incorporation of H_2 into blood by inhaling hydrogen gas. Rats inhaled a mixed gas of H_2 (1% or 2%) and O_2 (30%) under anesthetic O_2O and halothane for 1 h, and arterial (indicated by A) or venous blood (indicated by V) was collected into a closed aluminum bag from a three-way stopper (upper panel). After the transfer of O_2O into an accurate volume of air phase from the blood, amounts of O_2O were examined by gas chromatography. Lower panel shows profiles of gas chromatography. The vertical scale indicates the amounts of blood O_2O after calculations. Adapted from after Ohsawa et al. (2007) modified version of Fig. 5A, with permission from Nature Publishing Group.

current obtained from the H₂ sensor was converted to regional H₂ concentration using a calibration curve generated from known levels of H₂-saturated saline.



6. ADVANTAGES OF HYDROGEN IN MEDICAL APPLICATIONS

6.1. Selective reaction of H2 with highly reactive ROS

H₂ dissolved in culture medium did not change the cellular levels of ${}^{\bullet}O_2^-$ and H₂O₂, as judged by the fluorescent signals of MitoSOX and dichlorofluorescein-diacetate (DCF-DA), respectively (Ohsawa et al., 2007). Additionally, H₂ did not decrease the cellular level of NO ${}^{\bullet}$. In contrast, H₂ treatment significantly decreased levels of ${}^{\bullet}OH$, as judged by the decrease in the fluorescent signal of hydroxyphenyl fluorescein (HPF) (Setsukinai, Urano, Kakinuma, Majima, & Nagano, 2003).

In terms of the experimental protocol, culture media containing H_2 were prepared as follows: H2 was dissolved beyond the saturated level into DMEM medium under 0.4 MPa pressure of hydrogen gas for 2 h, O₂ was also dissolved into another medium by bubbling, the third medium contained CO₂, and then fetal bovine serum was supplemented to 1% in all three media. The three media were combined at various ratios to obtain the desired concentration of H₂ and 8.5 mg/l of O₂ at 25 °C. For culture, the combined media were put into a culture flask and immediately examined for H₂ or O₂ concentration with an H₂ or O₂ electrode, and in turn gas composed of the desired ratio of H_2 and N_2 ($H_2 + N_2 = 75\%$), 20% of O_2 , and 5% of CO₂ was filled into the culture flask; for example, when the medium contained 0.6 mM H₂, the H₂ gas was adjusted to 75%. The mixed gas was obtained by regulating the flow rates of its constituents with connected flow meters. As a control, degassed medium lacking H₂ was prepared by stirring medium that had been saturated with H₂ in an open vessel for 4 h, and the concentration of H₂ was checked with an H₂ electrode.

Then, PC12 cells were incubated in medium with or without 0.6 mM H₂, and exposed to antimycin A or L-NAME (N^G -nitro-L-arginine methyl ester) to induce ${}^{\bullet}O_2^-$, H₂O₂, and ${}^{\bullet}OH$ or NO ${}^{\bullet}$. Fluorescent images of MitoSOX-, DCF-DA (2',7'-dichlorodihydrofluorescein)-, HPF-, and DAF-2 DA (diaminofluorescein-2 diacetate)-treated cells were obtained by laser-scanning confocal microscopy (Olympus FV300) to estimate intracellular ${}^{\bullet}O_2^-$, H₂O₂, ${}^{\bullet}OH$, and NO, respectively (Fig. 2).

Alternatively, PC12 cells were exposed to intracellular ${}^{\bullet}OH$ produced by the Fenton reaction $(H_2O_2 + Cu^+ \rightarrow OH + OH^- + Cu^{2+})$, with or without 0.6 mM H_2 . Cells were preincubated with 1 mM CuSO₄, washed, and exposed for 1 h to 0.1 mM ascorbate (Vit. C) in order to reduce intracellular Cu $^{2+}$ to Cu $^+$. In this case, endogenous H_2O_2 would be sufficient to produce ${}^{\bullet}OH$. H_2 indeed protected the cells against ${}^{\bullet}OH$.

Moreover, the decrease in the cellular OH level by H₂ was confirmed by spin-trapping technology (Halliwell & Gutteridge, 1992). Standard electron spin resonance (ESR) signals of the DMPO-OH radical were obtained by trapping OH with a spin-trapping reagent (DMPO). PC12 cells were preincubated with 0.1 MDMPO and 2 mM CuSO₄ for 30 min at 37 °C with or without 0.6 mM H₂. After removal of this medium, the cells were treated with 0.2 mM ascorbate and 0.1 mM H₂O₂ for 5 min at 23 °C to produce OH by the Fenton reaction, and then scraped into a flat cuvette for ESR measurement. Alternatively, PC12 cells were incubated in PBS containing 0.1 M DMPO and 30 g/ml antimycin A for 7 min at 23 °C to produce

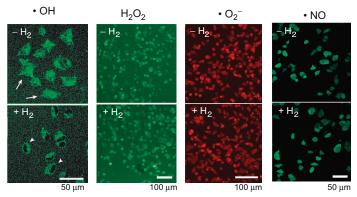


Figure 2 Selective reduction of reactive oxygen or nitrogen species by H_2 in cultured cells. PC12 cells were kept in medium with 0.6 m/M H_2 (indicated by $+H_2$) or without H_2 (indicated by $-H_2$), and exposed to antimycin A or L-NAME (N^G -nitro-L-arginine methyl ester) to induce 'OH, H_2O_2 , and ' O_2 ⁻ (ROS) or NO' (RNS) for 30 min. Each ROS or RNS was detected using flowing fluorescent dye; HPF, H_2DCF (2',7'-dichlorodihydrofluorescein), MitoSOX, and DAF-2 DA (diaminofluorescein-2 diacetate) were used to detect 'OH, H_2O_2 , ' O_2 ⁻, and NO', respectively. These representative fluorescence images obtained by laser-scanning confocal microscopy demonstrate the selective reduction of 'OH by H_2 . Adapted from Ohsawa et al. (2007) modified version of Fig. 1A, B and supplementary Fig. 1A, C, with permission from Nature Publishing Group.

excess ${}^{\bullet}OH$, with or without $0.6 \text{ m}MH_2$, and then scraped into a flat cuvette for ESR measurement.

The selective reduction of ROS can be explained by the marked oxidative strength of *OH. In other words, *OH is strong enough to react with even inert H₂, but that *O₂⁻, H₂O₂, and NO* are insufficient to react with H₂ according to their activities. Namely, H₂ is mild enough neither to disturb metabolic redox reactions nor to affect ROS that function in cellular signaling (Fig. 3).

6.2. Rapid diffusion

Most hydrophilic antioxidants cannot penetrate biomembranes and most hydrophobic antioxidants remain on the membranes. In contrast, H₂ can be infused into lipids as well as aqueous solutions. It has favorable distribution characteristics having the physical ability to penetrate biomembranes and diffuse into the cytosol, as illustrated in Fig. 4.

Despite the clinical importance of overcoming oxidative damage, antioxidants had limited therapeutic success. This may be because most

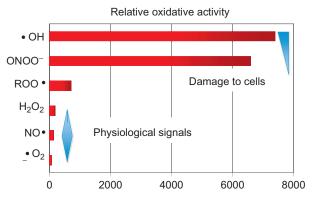


Figure 3 Relative oxidative activities in each reactive oxygen and nitrogen species. OH and ONOO are highly reactive to damaged cells, whereas O₂, NO, and H₂O₂ have physiological roles as signaling molecules. *This graph is based on data from a previous publication (Setsukinai et al., 2003).*

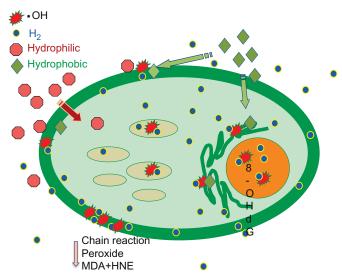


Figure 4 Illustration of gaseous diffusion of H_2 into a cell. Most hydrophilic compounds are retained at membranes and cannot reach the cytosol, whereas most hydrophobic ones cannot penetrate biomembranes in the absence of specific carriers. In contrast, H_2 can be rapidly distributed into cytosol and organelles. On the membrane, 'OH triggers the initiation of a free radical chain reaction to generate lipid peroxides, which are converted to some oxidative stress markers, 4-hydroxyl-2-nonenal (4-HNE), and malondialdehyde (MDA). In the nucleus, 'OH oxidizes DNA for modification to 8-OHdG (8-hydroxy-deoxyguanine).

antioxidants do not reach specific regions (Murphy, 1997; Murphy & Smith, 2000; Smith & Murphy, 2011). As H₂ effectively reaches the nucleus and mitochondria, the protection of nuclear DNA and mitochondria suggests preventive effects against lifestyle-related diseases, cancer, and the aging process (Ohsawa et al., 2007). Moreover, H₂ passes through the blood brain barrier, although most antioxidant compounds cannot; this is also an advantage of H₂.

The gaseous diffusion of H_2 can be monitored inside various tissues by detection with a specific H_2 electrode. For example, H_2 concentration has been monitored within the rat myocardium. The electrode was inserted into the "at-risk" area for infarction to estimate the diffusion of H_2 into the ischemic myocardium area after coronary artery occlusion. H_2 concentration was increased by its diffusion, even with coronary artery occlusion (Hayashida et al., 2008) (Fig. 5).

Moreover, we devised eye drops with dissolved H_2 to administer H_2 to the retina directly, and monitored the time course of changes in H_2 levels using the needle-shaped hydrogen sensor electrode inserted through the sclera to the vitreous body in rats. H_2 could reach the vitreous body by administering H_2 saturated in normal saline. When H_2 eye drops were

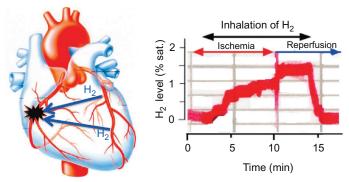


Figure 5 Inhalation of H_2 gas increases the intramyocardial H_2 by diffusion. Left panel illustrates that H_2 can reach an infarct area by diffusion even without blood flow. Right panel indicates an experimental result obtained as follows: Regional myocardial ischemia was induced by transient occlusion of the left anterior descending coronary artery of a rat. A needle-type hydrogen sensor (Unisense, Aarhus, Denmark) was inserted in the LV cavity (arterial blood) and H_2 gas at 2% was administered by respiration to intubated rats receiving mechanical ventilation and the concentration of H_2 in the "at-risk" area for infarction during ischemia and reperfusion was monitored with the needle-type H_2 sensor. Adapted from Hayashida et al. (2008) modified version of Fig. 2C, with permission from Elsevier.

administered continuously, approximately 70% H₂ was detected on the ocular surface (Oharazawa et al., 2010).

These experiments indicate that H₂ can rapidly diffuse into tissues even without blood flow.



7. METHODS OF INGESTING MOLECULAR HYDROGEN

7.1. Inhalation of hydrogen gas

Inhalation of H_2 gas is the most straightforward therapeutic method. H_2 gas can be inhaled through a ventilator circuit, facemask, or nasal cannula. Since inhaled H_2 gas acts rapidly, it may be suitable for defense against acute oxidative stress. In particular, inhalation of gas does not affect blood pressure (Ohsawa et al., 2007); on the other hand, drip infusion of drugs increases blood pressure and causes serious obstacles during the treatment of myocardial infarction. In particular, excess oxidative stress gives damages to tissues at the time of the initiation of reperfusion. Notably, most antioxidants cannot reach the at-risk area for infarction before initiating reperfusion. As pointed out above, H_2 can reach the region without blood flow by rapid diffusion (Fig. 5).

By a clinical examination, Ono et al. monitored H_2 and showed that inhalation of 3–4% H_2 gas did not affect any physiological parameters, suggesting no adverse effects (Ono et al., 2012).

7.2. Oral ingestion by drinking hydrogen water

Inhalation of H_2 gas is actually unsuitable or impractical for continuous H_2 consumption in daily life for preventive use. In contrast, solubilized H_2 (H_2 -dissolved water; i.e., H_2 -water) may be beneficial since it is a portable, easily administered, and safe way to ingest H_2 (Nagata, Nakashima-Kamimura, Mikami, Ohsawa, & Ohta, 2009; Ohsawa, Nishimaki, Yamagata, Ishikawa, & Ohta, 2008). H_2 can be dissolved in water up to 0.8 mM (1.6 mg/l) under atmospheric pressure at room temperature without any change of pH.

 H_2 -water can be made by several methods: infusing H_2 gas into water under high pressure, electrolyzing water to produce H_2 , and reacting magnesium metal or its hydride with water. These methods may be applicable not only to water but also to other solvents. H_2 penetrates glass and plastic walls of any vessel in a short time, while aluminum containers can retain H_2 for a long time.

In brief, for experimental treatments, H_2 was dissolved in water under high pressure (0.4 MPa) to a supersaturated level and the saturated H_2 -water was stored under atmospheric pressure in an aluminum bag with no dead volume. Mice were given water freely using closed glass vessels equipped with an outlet line containing two ball bearings, which kept the water from being degassed. The vessel was refilled with fresh H_2 -water at the same time (e.g., at 4:00 pm) every day.

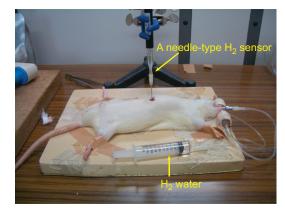
When water saturated with H₂ was placed into the stomach of a rat, H₂ was detected at several micromoles in blood (Nagata et al., 2009; Nakashima-Kamimura, Mori, Ohsawa, Asoh, & Ohta, 2009). In addition, a rat received H₂-water (0.8 mmol/l H₂ in water) orally by stomach gavage, for example, at 15 ml/kg. Hepatic H₂ was monitored with a needle-type hydrogen electrode (Kamimura, Nishimaki, Ohsawa, & Ohta, 2011) (Fig. 6).

Furthermore, after seven adult volunteers had drunk H₂-water, the H₂ content of their expired breath was measured by gas chromatography with a semiconductor (Shimouchi, Nose, Shirai, & Kondo, 2012). The ingestion of H₂-water rapidly increased breath H₂ content to its maximal level 10 min after ingestion, which thereafter decreased to the baseline level within 60 min. H₂ lost from the water during the experimental procedures accounted for 3% or less of the total. The rate of H₂ release from the skin surface was estimated as approximately 0.1%. On the basis of the remaining H₂ mass balance, approximately 40% of H₂ that had been drunk was consumed inside the body. This report suggests that exogenous H₂ is at least partially trapped by oxygen radicals, such as *OH (Shimouchi et al., 2012).

7.3. Injection of hydrogen-saline

 H_2 is intravenously or intraperitoneally injectable as H_2 -saline (H_2 -dissolved saline), which allows the delivery of H_2 with great efficacy in model animals (Cai et al., 2009; Li et al., 2013; Sun et al., 2011).

Nagatani et al. performed an open-label, prospective, nonrandomized study of intravenous H_2 administration in 38 patients hospitalized for acute ischemic stroke. All patients received an H_2 intravenous solution immediately after the diagnosis of acute ischemic stroke. Data from this study indicated that an H_2 intravenous solution is safe for patients with acute cerebral infarction, including patients treated with tissue-plasminogen activator (Nagatani et al., 2013).



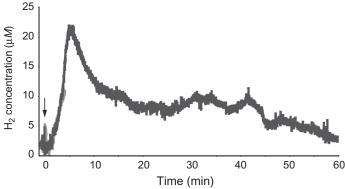


Figure 6 Incorporation of H_2 into the liver from the stomach. H_2 was dissolved in water under high pressure (0.4 MPa) to a supersaturated level. Upper panel: A needle-type hydrogen sensor (Unisense, Aarhus, Denmark) was inserted into rat liver and the rat received saturated H_2 -water orally by stomach gavage at 15 ml/kg (upper panel). Lower panel: H_2 concentration was monitored with a picoammeter (Keithley, Cleveland, Ohio). Arrow indicates the time point when rat was administered H_2 -water. Adapted from Kamimura et al. (2011), with permission from Wiley Online Library.

To rats, H₂-water, H₂-saline, and hydrogen gas were orally administered, intraperitoneally or intravenously injected, and inhaled, respectively. A method for determining the H₂ concentration was applied using high-quality sensor gas chromatography, after which the specimen was prepared via tissue homogenization in airtight tubes. The hydrogen concentration reached a peak at 5 min after oral and intraperitoneal administration, compared with 1 min after intravenous administration. These results indicate that H₂ can reach most organs or blood independently by the three methods (Liu et al., 2014).

7.4. Direct incorporation of molecular hydrogen by diffusion: Eye drops, bath, and cosmetics

Alternatively, H₂-loaded eye drops were prepared by dissolving H₂ in saline and directly administering them to the ocular surface (Kubota et al., 2011; Oharazawa et al., 2010).

 H_2 should easily penetrate the skin and is distributed throughout the body via blood flow. Thus, taking a warm water bath with dissolved H_2 is a method of incorporating H_2 into the body in daily life. It takes only 10 min for it to be distributed throughout the whole body, as judged by measuring H_2 gas in expiration (unpublished results). Indeed, powders that can be used to produce H_2 baths are commercially available in Japan.

 H_2 delivery to cardiac grafts during cold preservation using a hydrogensupplemented water bath efficiently ameliorated myocardial injury due to cold ischemia and reperfusion. This device to saturate organs with H_2 during cold storage merits further investigation for possible therapeutic and preventative use during transplantation (Noda et al., 2013).

7.5. Maternal intake of H₂

H₂ intake helps prevent the hippocampal impairment of offspring induced by ischemia/reperfusion during pregnancy (Mano et al., 2014). The effects of H₂ on rat fetal hippocampal damage caused by ischemia and reperfusion in pregnancy were examined with the transient occlusion of bilateral utero-ovarian arteries. Starting 2 days before the operation, the mothers were provided with H₂-saturated water *ad libitum* until vaginal delivery. A significant increase in the concentration of H₂ in the placenta was observed after the oral administration of H₂-saturated water to the mothers, with less placental oxidative damage after ischemia and reperfusion in the presence of H₂. Neonatal growth retardation was observed in the ischemia/reperfusion group, which was alleviated by H₂ administration. Maternal H₂ administration improved oxidative stress and the reference memory of the offspring to the sham level after ischemia and reperfusion injury during pregnancy. Thus, this finding supports the idea that maternal H₂ intake helps prevent the impairment of offspring induced by oxidative stress.



8. MEDICAL EFFECTS OF H₂

8.1. Acute oxidative stress by ischemia/reperfusion

As a type of acute oxidative stress, ischemia/reperfusion induces serious oxidative stress, and its injuries should be considered in many clinical treatments.

Inhalation of H₂ gas improved ischemia/reperfusion injuries in cerebral (Ohsawa et al., 2007) and myocardial infarction (Hayashida et al., 2008; Yoshida et al., 2012). Hydrogen-saline protected against renal ischemia/reperfusion injury (Wang et al., 2011). All clinical manifestations related to post-cardiac arrest (CA) syndrome are attributed to ischemia/reperfusion injury in various organs, including the brain and heart. H₂ gas inhalation yielded great improvement in survival and the neurological deficit score in post-CA syndrome in a rat model (Hayashida et al., 2012). H₂ also mitigated damage during the transplantation of various organs in the form of H₂ gas (Buchholz et al., 2008), H₂-water (Cardinal et al., 2010), and H₂-preservation solution (Noda et al., 2013). A clinical study showed a positive effect of H₂ on patients with acute brain stem infarction (Ono et al., 2011). These acute effects may be due to the direct reduction of oxidative stress by H₂ because no lag time was necessary.

8.2. Chronic oxidative stress loading to neurodegeneration

Chronic oxidative stress is accepted as one of the causes of neurodegeneration, including dementia and Parkinson's disease (PD) (Andersen, 2004; Federico et al., 2012). Experimental oxidative stress in the brain can be induced by chronic physical restraint stress and can impair learning and memory (Abrous, Koehl, & Le Moal, 2005; Liu et al., 1996). Drinking H₂-water suppressed the increase in this oxidative stress and prevented this cognitive impairment (Nagata et al., 2009). In PD, mitochondrial dysfunction and associated oxidative stress are major causes of dopaminergic cell loss in the substantia nigra (Schapira, 2008; Yoritaka et al., 1996). H₂ in drinking water was given before or after stereotactic surgery for 6-hydroxydopamine-induced nigrostriotal degeneration in a rat model of PD. H₂-water prevented both the development and the progression of nigrostriatal degeneration in rats (Fu et al., 2009). Moreover, drinking H₂-water also suppressed dopaminergic neuronal loss in another PD mouse model induced by MPTP (1methyl-4-phenyl-1,2,3,6-tetrahydropyridine) (Fujita et al., 2009). In a placebo-controlled, randomized, double-blind, parallel-group clinical pilot study, the efficacy of H₂-water in patients with PD was assessed for 48 weeks. Total Unified Parkinson's Disease Rating Scale (UPDRS) scores in the H₂water group significantly improved, whereas UPDRS scores in the placebo group worsened (Yoritaka et al., 2013).

8.3. Stimulatory effects on energy metabolism

Obesity induces oxidative stress (Matsuda & Shimomura, 2013). H_2 -water significantly alleviated fatty liver in db/db mice, which are type 2 diabetes model mice with obesity, as well as high-fat diet-induced fatty liver in

wild-type mice. Long-term H₂-water drinking significantly decreased fat and body weights, despite no increase in the consumption of diet and water, in *db/db* mice, and decreased levels of plasma glucose, insulin, and triglyceride by stimulating energy metabolism (Kamimura et al., 2011). Analysis of gene expression revealed that a hepatic hormone, fibroblast growth factor 21 (FGF21), showed increased expression upon drinking H₂-water (Kamimura et al., 2011). FGF21 functions to stimulate fatty acid and glucose expenditure. Thus, H₂-water stimulates energy metabolism (Kamimura et al., 2011). Beneficial roles of H₂-water in the prevention of potential metabolic syndrome were also reported by a clinical study (Song et al., 2013).

8.4. Anti-inflammatory effects

Inflammation is closely involved in oxidative stress. H₂-reduced inflammation in experimental model animals induced by concanavalin A and dextran sodium sulfate (Kajiya, Silva, Sato, Ouhara, & Kawai, 2009), lipopolysaccharide (Chen et al., 2013; Xu et al., 2012), Zymosan, an inducer of generalized inflammation and polymicrobial sepsis (Li et al., 2013). H₂ gas, H₂-saline, and H₂-water decreased the levels of proinflammatory cytokines to suppress inflammation. Rheumatoid arthritis (RA) is a chronic inflammatory disease characterized by the destruction of bone and cartilage. The symptoms of RA were significantly improved with H₂-water (Ishibashi et al., 2012).

In terms of the current state of knowledge, H₂ exhibits not only antioxidative effects but also affects many phenotypes in various model animals. H₂ has many beneficial effects on animal models and patients besides its antioxidative effects: anti-inflammation, antiapoptosis, antiallergy, and stimulation of energy metabolism (Ohta, 2011, 2012, 2014). Their mutual relationships are not clear, but the reduction of oxidative stress may primarily lead to various subsequent effects. H₂ seems to exhibit a variety of phenotypic effects toward improving many pathogenic states by regulating the expression of various genes. The molecules encoding by these genes are, probably, not primary responders to H₂, but indirectly act to enable the various effects of H₂. The primary target of H₂ remains unknown.



9. POSSIBLE MOLECULAR MECHANISMS UNDERLYING VARIOUS EFFECTS OF MOLECULAR HYDROGEN

9.1. Direct reduction of hydroxyl radicals with molecular hydrogen

H₂ was shown to reduce OH in an experiment using cultured cells (Ohsawa et al., 2007). Later, it was shown that H₂ eye drops directly decreased OH

induced by ischemia/reperfusion in retinas (Oharazawa et al., 2010). Moreover, it has been demonstrated that, at the tissue level, H₂ neutralized *OH that had been induced by ionizing irradiation in testes, as judged by the decreased HFP signal, and exhibited a radioprotective role (Chuai et al., 2012).

Considering the reaction rate of OH with H2 in dilute aqueous solutions, this rate may be too slow to enable fully a decrease in OH in order to exhibit its beneficial roles (Buxton, Greenstock, Helman, & Ross, 1988). Mammalian cells are, however, highly structured with complicated biomembranes and viscous solutions with multiple concentrated components. Since collision frequency is rate-limiting in a viscous environment, the marked diffusion rate of H2 could be advantageous to overcome the slow reaction rate constant. OH is known as a major trigger of the chain reaction of free radicals (Niki, 2009). Once this chain reaction occurs on biomembranes, it continues and expands causing serious damage to cells (Fig. 4). H₂ accumulates in the lipid phase more than in the aqueous phase, especially in unsaturated lipid regions, which are the major target of the initial chain reaction (unpublished results). Thus, H₂ may have an advantage to suppress the chain reaction, which produces lipid peroxide, and leads to the generation of oxidative stress markers, such as 4-hydroxyl-2-nonenal (4-HNE) and malondialdehyde (MDA) (Niki, 2014). Indeed, H₂ decreased these oxidative markers in many studies (Ning et al., 2013; Ohsawa et al., 2008; Zhou et al., 2013). Additionally, OH can modify deoxyguanine (dG) to 8-hydroxy-deoxyguanine (8-OHdG) (Delaney, Jarem, Volle, & Yennie, 2012; Kawai et al., 2012) (Fig. 4). H₂ decreased the level of 8-OHdG in most of the examined patients and animals (Ishibashi et al., 2012; Kawai et al., 2012).

These experimental observations suggest that sufficient H_2 can efficiently mitigate tissue oxidation induced by *OH. However, when animals or humans drink H_2 -water, it is not clear whether H_2 -water provides a sufficient amount of H_2 to scavenge *OH efficiently (Fig. 7).

9.2. Direct reduction of peroxynitrite with molecular hydrogen to regulate gene expression

As another molecular mechanism, the scavenging of ONOO⁻ by H₂ should be considered. ONOO⁻ is known to modify tyrosine of proteins to generate nitro-tyrosine (Radi, 2013). Several studies have shown that H₂ efficiently decreases nitro-tyrosine in animal models regardless of whether H₂-water (Cardinal et al., 2010), H₂ gas (Shinbo et al., 2013), or H₂-saline

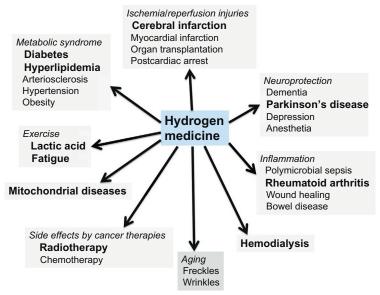


Figure 7 Summary of potential of various preventive and therapeutic effects of H₂. Bold letters indicate published results for clinical examinations (Ohta, 2014). Positive effects obtained by animal experiments for disease models are shown by normal text.

(Chen et al., 2010; Yu et al., 2011; Zhang et al., 2011; Zhu et al., 2011) is used. Moreover, drinking H_2 -water decreased nitro-tyrosine in patients with RA (Ishibashi et al., 2012). Thus, at least part of the effect of H_2 can be attributed to the decreased production of nitro-tyrosine in proteins.

Many protein factors involved in transcriptional control are nitrolated (-O-NO₂) or nitrosolated (-S-NO₂). It is possible that the decrease in -O-NO₂ or -S-NO₂ may regulate the expression of various genes (Radi, 2013). However, major targets have not been identified and are under investigation.

9.3. Indirect reduction of oxidative stress by regulating gene expression

H₂ reduces oxidative stress not only directly, but also indirectly, by inducing antioxidation systems, including HO-1 SOD (Zhai et al., 2013), catalase (Cai, Zhang, Yu, & Cai, 2013), and myeloperoxidase (Zhang et al., 2011). Nrf2 is known to function as a defense system against oxidative stress and various poisons by inducing various genes including HO-1. HO-1, a microsomal enzyme degrading heme to carbon monoxide, free iron, and

biliverdin, participates in the cell defense against oxidative stress (Jazwa & Cuadrado, 2010).

In Nrf2-deficient mice, mitigating effects by the inhalation of H₂ gas declined in hyperoxic lung injury accompanying by a decrease in HO-1, indicating that H₂ gas can ameliorate hyperoxic lung injury in an Nrf2-dependent manner (Kawamura et al., 2013). Activation of Nrf2 is also required for the amelioration of cerebral ischemia–reperfusion injury in rats by H₂ (Zhai et al., 2013).

 H_2 influences some signal transductions as an indirect modulator; however, it is unlikely that H_2 could directly bind to some receptors involved in the signal transductions. The primary target molecule of H_2 has not been identified in these signal transduction pathways. These regulatory molecules are, probably, not primary responders to H_2 , but indirectly act to enable the various effects of H_2 . The primary target of H_2 remains unknown.

10. UNRESOLVED QUESTIONS AND CLOSING REMARKS

 H_2 can be incorporated or ingested into the body by various methods: inhalation of H₂ gas, drinking H₂-infused water (H₂-water), injection of H₂infused saline, and incorporation through the skin. Drinking H₂-water was efficacious for various disease models and patients; however, H_2 can be infused up to only 0.8 mM under atmospheric pressure and drinking H₂water provides a blood H₂ concentration up to only $\sim 10 \,\mu M$ with short dwelling time in the body (Nagata et al., 2009; Nakashima-Kamimura et al., 2009). Moreover, inhaling 1-4% (vol/vol) of H₂ gas was effective, by which H_2 should reach 8-32 μM in blood. Under these conditions, H₂ should be insufficient to scavenge OH for fully exhibiting H₂ benefits because the direct reaction rate of OH with H₂ in an aqueous solution may be too slow to decrease OH (Buxton et al., 1988) as pointed out earlier (Wood & Gladwin, 2007). Thus, it remained elusive how such low levels of H₂ with a short dwelling time could effectively compete with the numerous cellular targets in chronic or acute pathogenesis. Unexpectedly, H₂ was shown to regulate the expression of many genes and the phosphorylation of factors involved in various types of signal transduction to exhibit various phenotypes. For example, drinking H₂-water reduces the gene expressions of proinflammatory cytokines to relieve inflammation, as mentioned above, FGF21 to stimulate energy metabolism (Kamimura et al., 2011), and Grelin for neuroprotection (Matsumoto et al., 2013). However, it essentially remains unsolved what the primary target of H₂ is.

Many other mysteries regarding H_2 therapy also remain unresolved. For initiating cellular signals by H_2 , H_2 should be too inert to react with most molecules except highly reactive ones, such as 'OH or ONOO'. To activate H_2 to react with other molecules, a sufficient level of a putative catalyst must be present; however, it is highly unlikely that such a putative catalyst would be abundant. Moreover, H_2 should be too small to bind a putative H_2 -binding receptor because its intramolecular fluctuation should lead to the instability.

H₂ can be easily applied because of a lack of adverse effects and great efficacy for nearly all pathogenic statuses involved in oxidative stress and inflammation. Since most pharmacological drugs specifically act on their targets, H₂ seems to differ from conventional drugs or other medical gasses because of its extensive and varied effects. H₂ has great potential for preventive and therapeutic applications owing to its great efficacy and its "novel" concept.

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